Course Information									
Course Code	Т	P	L	C	ECTS	Type C/E	Language TR/ENG etc.	Year/Semester	
KİM1013	0	0	2	1	2	C	TR	1/FALL	
Course Name (Turkish)	Genel Ki	Genel Kimya Laboratuvarı-I							
Course Name (English)	General	General Chemistry Laboratory-I							

Unit/Program	Physics Department/Undergraduate Program									
Course Prerequisite	No									
Course Objectives	The aim of this course is to give information about the basic experimental studies and laboratory techniques used in chemistry.									
Course Outline	Basic Chemistry Exp	Basic Chemistry Experiments								
Textbook/ Material / Resources	General Chemistry I	General Chemistry Laboratory Experiments Test Sheet								
Internship Status	Internship									
	Course Precedents									
University Name	Program Name	Course Name	T-P-L-C; ECTS	Туре						
Izmir University of Technology	Physics	General Chemistry Laboratory I	0-0-2-1, 2	С						
Balıkesir University	Physics	General Chemistry Laboratory I	0-0-2-1, 2	C						
Eskisehir Osmangazi University	Physics	General Chemistry Laboratory I	0-0-2-1, 2	C						
The instructor wh	o proposed the cours	se (Title, Name and Surname)	Signatu	ıre						
Instructors who c	an teach the course (Title, Name and Surname)	Signatu	ıre						

Academic justification for the opening of the course? (The effect of course outcomes on program outcomes, etc.)

Brief explanation of the course (theoretical lecture, applications, laboratory, studio, off-campus activity, using software, etc.)

External Stakeholder Opinions About the Course (It is expected that the opinions to be obtained from the business								
world that will employ your graduates or from real or legal persons outside the University who have expertise on the subject of								
the course will be specified. Proof documents mu	st be attached to this form.)							
Stakeholder Name	Opinion (It should be given as a summary, it should not exceed two lines.)							

	Weekly Course Content Distribution								
Week	Theory	Application/Laboratory							
1	Basic Laboratory Principles								
2	Safety in the Laboratory								
3	Accidents that may occur in the laboratory and how to respond to them								
4		Laboratory Supplies							
5		Volume Measurement and Weighing Process in the Laboratory							
6		Cleaning Glass Materials							
7		Preparation of Washing Solutions							
8		Basic Physical and Chemical Processes							
9	Midterm Exam								
10		Heating and cooling							
11		Law of Conservation of Mass							
12		Separation Methods							
13		Solutions							
14		Equivalent Weight of Magnesium							
15	Finale								
16									

Assessment							
	Activity	Custom	Contribution to Success Grade (%)				
Evaluation Criteria	Midterm Exams	1	40				
	Quizzes						
	Assignments						
	Projects						
	Term Paper						
	Laboratory						
	Other						
	Final Exam	1	60				
		Sum:	100				
Remarks		-					
	Mathematics and Basic	100					

Content Design and Subject Weight (%)	Mathematics and Basic Sciences	100
	Engineering Sciences	
	Social Sciences	
	Health Sciences	
	Educational Sciences	
	Culture and Art Sciences	
	Design Information	

Workload (ECTS) Calculation							
Events	Number	Duration (Hours)	Total workload (Hours)				
Fieldwork							
Midterm Exam Application	1	1	1				
Self-Study (including pre-class and exam preparation)	3	8	24				
Make-up Exam	1	1	1				
Experiment and Observation		_	-				
Class Participation (Theory)							
Homework							
Final Exam Practice	1	1	1				
Laboratory	14	2	28				
Article Review							
Writing an Article							
Reading							
Case Study							
Performance							
Problem Solution							
Project Preparation							
Project Submission							
Quiz							
Report Preparation							
Submitting Reports							
Role/Drama Work							
Seminar							
Oral Exam							
Team/Group Work							
Argument							
Application/Practice							
Other							
	Т	OTAL WORKLOAD:	55				
(The number obtained as a result of Total	ECTS CREDITS OF THE COURSE: (The number obtained as a result of Total Workload/25 is calculated by rounding to the whole number.)						

	The Relationship Between Course Learning Outcomes and Program Outcomes											
I	Program Outcomes (PO) Learning Outcomes (LO) (Course Outcomes)	1	2	3	4	5	6	7	8	9	10	11
1	Comprehension of the fundamentals of chemistry	1	1	1	1	1	1	5	5	5	1	1
2	Ability to apply basic knowledge of Chemistry, Mathematics and Physics to Chemistry problems	5	5	5	5	3	1	5	5	5	1	1
3	Ability to identify, define, analyze and solve problems in chemistry and related fields	3	3	3	5	3	1	5	5	5	1	1

Organizer: Assoc. Prof. Dr. Ersin PEKDEMİR

Preparation Date: 20.05.2024