Course Information										
Course Code	Т	Р	L	С	ECTS	Type C/E	Language TR/ENG etc.	Year/Semester		
KİM1014	0	0	2	1	2	С	TR	1/SPRING		
Course Name (Turkish)	me sh) Genel Kimya Laboratuvarı-II									
Course Name (English)	General	General Chemistry Laboratory-II								

Unit/Program	Physics Departme	Physics Department/Undergraduate Program							
Course Prerequisite	No	No							
Course Objectives	The aim of this con laboratory technique	he aim of this course is to give information about the basic experimental studies and aboratory techniques used in chemistry.							
Course Outline	Basic Chemistry Exp	Basic Chemistry Experiments							
Textbook/ Material / Resources	Textbook/ Material / General Chemistry Laboratory Experiments Test Sheet Resources								
Internship Status									
	Course Precedents								
University Name	Program Name	Course Name	T-P-L-C; ECTS	Туре					
Izmir University of Technology	Physics	General Chemistry Laboratory II	0-0-2-1, 2	С					
Balıkesir University	Physics	General Chemistry Laboratory II	0-0-2-1, 2	С					
Eskisehir Osmangazi University	Physics	General Chemistry Laboratory II	0-0-2-1, 2	С					
The instructor wh	Signature								
Instructors who can teach the course (Title, Name and Surname)				е					

Academic justification for the opening of the course? (The effect of course outcomes on program outcomes, etc.)

Brief explanation of the course (theoretical lecture, applications, laboratory, studio, off-campus activity, using software, etc.)

External Stakeholder Opinions About the Course (It is expected that the opinions to be obtained from the business							
world that will employ your gradu	world that will employ your graduates or from real or legal persons outside the University who have expertise on the subject of						
the course will be specified. Proof	documents must be attached to this form.)						
Stakeholder Name Opinion (It should be given as a summary, it should not exceed two lines.)							
Weekly Course Content Distribution							

Week	Theory	Application/Laboratory			
1	Basic Laboratory Principles				
2	Safety in the Laboratory				
3		Determination of Atomic Weight			
4		Determination of Crystal Water			
5		Investigation of Diffusion Phenomenon in Gases			
6		Catalysts			
7		pH and Indicators			
8		Acid-Base Titrations			
9	Midterm Exam				
10		Preparation and Adjustment of Acid-Base Solutions			
11		Buffer Solutions			
12		Sublimation			
13		Determination of Molecular Weight by Freezing Point Descent			
14		Determination of Elements in Organic Compounds			
15	Finale				
16					

	Assessment		
	Activity	Contribution to Success Grade (%)	
	Midterm Exams	1	40
	Quizzes		
	Assignments		
Evaluation Criteria	Projects		
	Term Paper		
	Laboratory		
	Other		
	Final Exam 1		60
		Sum:	100
Remarks			

Content Design and	Mathematics and Basic Sciences	100
	Engineering Sciences	
	Social Sciences	
Subject Weight	Health Sciences	
(%)	Educational Sciences	
	Culture and Art Sciences	
	Design Information	

Workload (ECTS) Calculation							
Events	Number	Duration (Hours)	Total workload (Hours)				
Fieldwork							
Midterm Exam Application	1	1	1				
Self-Study (including pre-class and exam preparation)	3	8	24				
Make-up Exam	1	1	1				
Experiment and Observation							
Class Participation (Theory)							
Homework							
Final Exam Practice	1	1	1				
Laboratory	14	2	28				
Article Review							
Writing an Article							
Reading							
Case Study							
Performance							
Problem Solution							
Project Preparation							
Project Submission							
Quiz							
Report Preparation							
Submitting Reports							
Role/Drama Work							
Seminar							
Oral Exam							
Team/Group Work							
Argument							
Application/Practice							
Other							
	T	OTAL WORKLOAD:	55				
EC (The number obtained as a result of Total ro	2						

	The Relationship Between Course Learning Outcomes and Program Outcomes											
L	Program Outcomes (PO) earning Outcomes (LO) (Course Outcomes)	1	2	3	4	5	6	7	8	9	10	11
1	Comprehension of the fundamentals of chemistry	1	1	1	1	1	1	5	5	5	1	1
2	Ability to apply basic knowledge of Chemistry, Mathematics and Physics to Chemistry problems	5	5	5	5	3	1	5	5	5	1	1
3	Ability to identify, define, analyze and solve problems in chemistry and related fields	3	3	3	5	3	1	5	5	5	1	1
4	Ability to use techniques, methods and modern tools required for chemistry applications	3	3	3	5	3	1	5	5	5	1	1

Organizer: Assoc. Prof. Dr. Ersin PEKDEMİR **Preparation Date:** 20.05.2024