Course Information										
Course Code	Т	Р	L	С	ECTS	Type C/E	Language TR/ENG etc.	Year/Semester		
FİZ3006	4	2 0 5 5		5	С	TR	3/Spring			
Course Name (Turkish)	Atom ve	Atom ve Molekül Fiziği Atom and Molecules Physics								
Course Name (English)	Atom an									

Unit/Program	Physics Department/Undergraduate Program									
Course Prerequisite	No									
Course Objectives	To study the structure of atoms and molecules, their interactions with each other and their dynamics. In addition, to explain the macroscopic properties of matter on the basis of atoms, which are microscopic components of matter, and to define the relationships between macroscopic and microscopic properties of matter.									
Course Outline	Orbital and spin motions in o spectroscopy, Molecular Stru Molecules, Molecular spectro	Orbital and spin motions in one-electron atoms, Multi-electron atoms and atomic spectroscopy, Molecular Structure, Electronics, Vibration and Rotational Motions of Molecules, Molecular spectroscopy								
Textbook/ Material / Resources	1.Atomic and Molecular Physics: Translation: Fevzi KÖKSAL and Hasan GÜMÜŞ 2.Concepts of Modern Physics: Translation: Gülsen ÖNENGÜT									
Internship Status	No									
	С	ourse Precedents								
University Name	Program Name	T-U-L-K; AKTS	Туре							
The instructor wh	o proposed the course (Title,	Signature								
Instructors who c	an teach the course (Title, Nar	ne and Surname)	Signature							

Academic justification for the opening of the course? (The effect of course outcomes on program outcomes, etc.)

Brief explanation of the course (theoretical lecture, applications, laboratory, studio, off-campus activity, using software, etc.)

The lesson with the students will be taught face-to-face on the board.

External Stakeholder Opinions About the Course (It is expected that the opinions to be obtained from the business							
world that will employ your graduates or from real or legal persons outside the University who have expertise on the subject of							
the course will be specified. Proof docume	ents must be attached to this form.)						
Stakeholder Name Opinion (It should be given as a summary, it should not exceed two lines.)							

	Weekly Course Content Distribution	
Week	Theory	Application/Laboratory
1	Atom Modelleri, Bohr Atom Models	
2	Quantum Theory of Atoms with One Electron, Orbital Motion	
3	Quantum Theory of Atoms with One Electron, Spin Motion of Electron	
4	Thin and Extremely Thin Structures	
5	Zeeman and Stark Incidents	
6	Possibility of passage and Selection Rules	
7	Spectroscopic Notations and Term symbols	
8	Multi-electron Atoms, Independent Particle Model and Variation Principle	
9	Midterm Exam	
10	Pauli's principle, Hund's Rules, and term symbols for multi-electron atoms	
11	Molecular Structure and Types of Molecular Bonds	
12	Electronic Schrödinger Equation and Molecular Orbitals	
13	Vibrational Motions of Molecular and Vibrational Spectroscopy	
14	Rotational Motions of Molecules and Rotational Spectroscopy	
15	Finale	
16		

	Assessment		
	Activity	Custom	Contribution to Success Grade (%)
	Midterm Exams	1	%40
	Quizzes		
	Assignments		
Evaluation Criteria	Projects		
	Term Paper		
	Laboratory		
	Other		
	Final Exam	1	%60
		Sum:	100
Remarks			

	Mathematics and Basic Sciences	%100
	Engineering Sciences	
Content Design and	Social Sciences	
Subject Weight	Health Sciences	
(%)	Educational Sciences	
	Culture and Art Sciences	
	Design Information	

Workload (
Events	Number	Duration (Hours)	Total workload (Hours)
Fieldwork			
Midterm Exam Application	1	2	2
Self-Study (including pre-class and exam preparation)			
Make-up Exam	1	2	2
Experiment and Observation			
Class Participation (Theory)	14	6	84
Homework		-	
Final Exam Practice	1	2	2
Laboratory			
Article Review			
Writing an Article			
Reading			
Case Study			
Performance			
Problem Solution	14	2	28
Project Preparation			
Project Submission			
Quiz			
Report Preparation			
Submitting Reports			
Role/Drama Work			
Seminar			
Oral Exam			
Team/Group Work			
Argument			
Application/Practice	7	1	7
Other			
	125		
EC (The number obtained as a result of Total ro	5		

	The Relationship Between Course Learning Outcomes and Program Outcomes												
I	Program Outcomes (PO) earning Outcomes (LO) (Course Outcomes)	1	2	3	4	5	6	7	8	9	10	11	12
1	To be able to explain the microscopic structure of substances consisting of different atoms and molecules	5	5	5	3	3	1	5	4	5	1	1	
2	To be able to distinguish and define the difference between classical and quantum physics	5	5	5	3	3	1	5	4	5	1	1	
3	To be able to explain the basic structures of hydrogen and helium atoms using Ouantum Physics theories	5	5	5	3	3	1	5	4	5	1	1	

Organizer: Assoc. Prof. Dr. Seda HEKİM Preparation Date: 20.05.2024